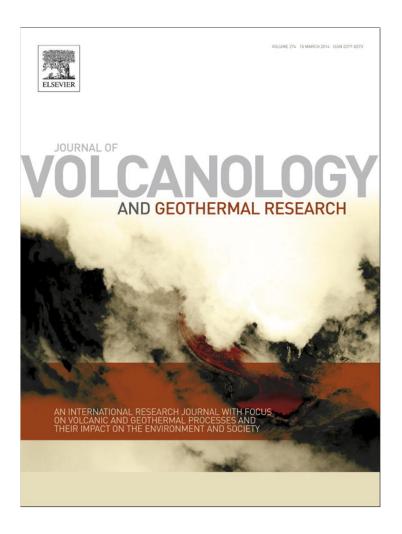
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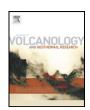
Journal of Volcanology and Geothermal Research 274 (2014) 71–77



Contents lists available at ScienceDirect

Journal of Volcanology and Geothermal Research

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The Domuyo volcanic system: An enormous geothermal resource in Argentine Patagonia



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ARTICLE INFO

Article history: Received 24 November 2013 Accepted 5 February 2014 Available online 15 February 2014

Keywords: Domuyo volcano Argentine Patagonia Geothermal potential Water geochemistry

ABSTRACT

A geochemical survey of the main thermal waters discharging in the southwestern part of the Domuyo volcanic complex (Argentina), where the latest volcanic activity dates to 0.11 Ma, has highlighted the extraordinarily high heat loss from this remote site in Patagonia. The thermal water discharges are mostly Na-Cl in composition and have TDS values up to 3.78 g L $^{-1}$ (El Humazo). A simple hydrogeochemical approach shows that 1,100 to 1,300 kg s $^{-1}$ of boiling waters, which have been affected by shallow steam separation, flow into the main drainage of the area (Rio Varvarco). A dramatic increase of the most conservative species such as Na, Cl and Li from the Rio Varvarco from upstream to downstream was observed and related solely to the contribution of hydrothermal fluids. The equilibrium temperatures of the discharging thermal fluids, calculated on the basis of the Na-K-Mg geothermometer, are between 190 °C and 230 °C. If we refer to a liquid originally at 220 °C (enthalpy = 944 J g $^{-1}$), the thermal energy release can be estimated as high as 1.1 \pm 0.2 GW, a value that is much higher than the natural release of heat in other important geothermal fields worldwide, e.g., Mutnovsky (Russia), Wairakei (New Zealand) and Lassen Peak (USA). This value is the second highest measured advective heat flux from any hydrothermal system on Earth after Yellowstone.

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1. Introduction

In tectonically young and active areas of the Earth, where meteoric waters infiltrate and circulate deeply, advective heat flux may be the dominant form of heat transfer and should be considered in the evaluation of the geothermal heat flux (e.g., Bodmer and Rybach, 1985; Čermak and Jetel, 1985; Gosnold, 1990; Bodri and Rybach, 1998; Brumm et al., 2009; Ingebritsen and Mariner, 2010; Chiodini et al., 2013). The magnitude and spatial distribution of heat flux are relevant for evaluating potential geothermal resources and also for inferring thermal the deep structure underlying volcanic areas, with reference to the nature and size of heat sources (such as magma or hot intrusive rocks) at depth (Manga, 1998). The fluid discharge rates of major thermal areas worldwide range from a few liters to several cubic meters of hot water per second (Taran and Peiffer, 2009). At Yellowstone Caldera (USA), the total discharge of the thermal waters was estimated to be

3,200 L s⁻¹ (Fournier, 1989). However, thermal discharges in other geothermal systems are rarely higher than 100 L s⁻¹, this being for instance the reference value computed for pre-exploitation discharge at Wairakei (New Zealand; Ellis and Wilson, 1955). Thermal water flow rates of about 80 L s⁻¹ were estimated at the Mutnovsky geothermal field in Kamchatka, Russia (Vakin and Pilipenko, 1986), while about 20 L s⁻¹ of thermal water flow was measured at the Lassen Peak hydrothermal system (USA; Sorey, 1986). Others authors (Manga, 1998) stressed that cold or "slightly thermal" springs can also strongly affect the heat budget of a thermal zone, showing that in the central Oregon Cascades roughly half of the advective geothermal heat flux is discharged from springs only a few degrees above local ambient temperature.

In this work, we present the results of a geochemical survey carried out on boiling water discharges located at the remote volcano of Domuyo, a scarcely investigated volcanic complex from Argentine Patagonia (Fig. 1). Palacio and Llambias (1978) provided the first geochemical characterization of the Domuyo waters while other chemical data were reported in two internal reports produced by JICA (1983; 1984) in the framework of a joint collaboration between the Government of

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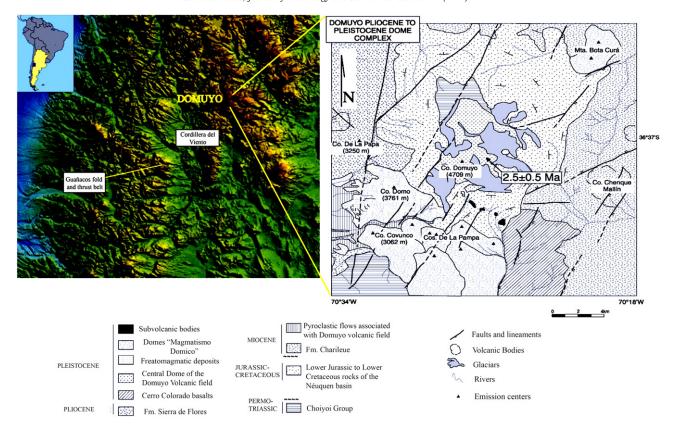


Fig. 1. (Left panel) The Domuyo volcanic complex, north of the Cordillera del Viento chain, and (right panel) geological map showing the distribution of Permian–Triassic to Pleistocene rocks; the arrow points to a K-Ar dating location (modified from Miranda et al., 2006).

Argentina and the Government of Japan to evaluate the geothermal resources in the area.

On the western slope of the Domuyo volcano, several boiling thermal springs discharge into roughly ENE-WSW-oriented creeks, which in turn flow into the main drainage system of the southwestern part of the volcanic edifice: the N-S-oriented Rio Varvarco (Fig. 2). A geochemical approach similar to those successfully applied to Yellowstone (Fournier, 1989) and El Chichon (Mexico, Taran and Peiffer, 2009) was performed in this sector of the Domuyo volcano in order to estimate its thermal release and geothermal potential. The thermal energy was computed by (i) evaluating the flux of chloride from the thermal source into the rivers and (ii) determining the original enthalpy and chloride content of the geothermal brines. Our results demonstrate that the thermal discharges account for a surprisingly high-energy release of 1.1 ± 0.2 GW.

2. Regional setting

2.1. Geological setting and volcanic activity

The Domuyo volcanic complex (36°340 S, 70°250 W, 4,709 m high), nicknamed the "Roof of Patagonia," is Middle Miocene to Early Pliocene in age and is located at the northern edge of the Cordillera del Viento (Fig. 1), a chain uplifted during Cretaceous and Neogene times (Kay et al., 2006) that, together with the late Miocene to Quaternary Guañacos fold and thrust-belt mountain system (Suárez and Emparan, 1995, 1997) to its west, constitutes the main geological feature of the study area. At larger scale, the Domuyo volcano belongs to the southern volcanic zone (SVZ: 33°S to 46°S), which has evolved for the last 20 Ma years by slightly dextral-oblique convergence between the Nazca and the South America plates at a rate of ca. 7–9 cm/year (Cembrano and Lara, 2009 and references therein).

The Domuyo volcano, with an estimated volume of ~ 200 km³ and a base area of 138 km² (Völker et al., 2011), is a broad structural dome and forms, together with three other tectonic domes, a major anticline, the "La Cruzada" high (Folguera et al., 2007). The geology of the area, initially studied by Groeber (1947), is made up of Permian-Triassic sedimentary and pyroclastic rocks (Choyoi Group) and acidic intrusive rocks, the latter intruding older formations. Jurassic and Cretaceous sedimentary and pyroclastic rocks unconformably overlie the basement (Spagnuolo et al., 2012). Tertiary andesitic volcanism covered the Mesozoic rocks and intrusive bodies emplaced close to the Domuyo volcano. Rhyolitic to dacitic tuff, lapilli tuff and lava flows cover its western slope. According to Mas et al. (2009), intrusive stocks are centered on the volcano and outcrop in an area of about 24 km² at Domuyo, intruding the older sedimentary cover. Volcanism resumed in the late Pliocene to Quaternary during which time sub-volcanic bodies (K/Ar 2.5 \pm 0.5 Ma; Miranda, 1996, 2006) were also emplaced. The volcanic rocks surrounding the Domuyo volcano belong to two distinct magmatic series, originating from two different feeding magmas (Brousse and Pesce, 1982). One of these (Late Pliocene-Early Pleistocene) is calcalkaline and yielded andesitic rocks; the other (Middle Pleistocene) is represented by both shoshonitic and alkaline series and has formed dacitic and/or rhyolitic rocks of which the main expression is the Cerro Domo (absolute age 0.72 \pm 10 Ma calculated on a rhyolitic lava flow; Brousse and Pesce, 1982), located on the southern slope of Domuyo. Younger volcanic products were dated between 0.55 \pm 0.10 and 0.11 \pm 0.02 Ma (JICA, 1983) by the fission-track method, suggesting that Domuyo volcanism has been active through the Late Pleistocene.

2.2. Hydrothermal activity

This remote site in Patagonia has impressive signs and obvious evidence of high release of thermal energy. High flow rates (hundreds

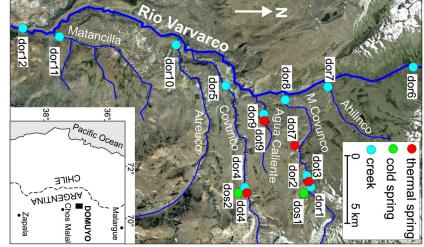


Fig. 2. Location of sampling sites in the SW slope of Domuyo volcano.

Table 1 Geographical coordinates, altitudes, flow rates, physico-chemical features and geochemical facies of the studied waters.

 $L\,s^{-1})$ of boiling fluids (Table 1, Fig. 3) enter the hydrographic network, which are able to maintain the temperature of the creeks up to 30 °C for

of about 300 m was visible up to a distance of 20 km (Mas et al., 2009)

Sample	Locality	Type	Latitude	Longitude	Altitude	T	pH	EC	Flow rate	Na	K	Mg	Ca	Li	Cl	HCO_3	SO_4	TDS	Composition
			UTM	UTM	m	°C		mS/cm	kg s ⁻¹				mg kg	1				g kg ⁻¹	
dor1	Manchana Covunco	Creek	361029	5942745	1,995	21.2	7.96	1,641	nd	169	17.8	7.74	113.0	na	245	115	129	0.80	Na-Cl
dor2	Manchana Covunco	Creek	360826	5942649	1,952	48.5	8.01	3,860	800	526	53.8	3.04	92.9	5.08	799	100	257	1.84	Na-Cl
dor4	Covunco	Creek	361863	5938249	2,038	37.2	na	3,100	840	462	44.4	3.30	168.3	4.23	722	84	220	1.71	Na-Cl
dor5	Covunco	Creek	354135	5936607	1,541	23.1	8.15	2,506	nd	404	36.6	2.38	64.8	3.60	629	74	169	1.38	Na-Cl
dor6	Rio Varvarco	Rio	352305	5950486	1,581	10.5	8.08	416	6,630	5.98	1.4	2.54	79.1	0.003	1.8	56	145	0.29	Ca-SO ₄
dor7	Ahilinco	Creek	356191	5945057	1,658	11.5	8.39	147	nd	14.9	2.5	4.89	10.9	na	17.2	55	1.02	0.11	Ca-HCO ₃
dor8	Manchana Covunco	Creek	355626	5942334	1,585	29.7	8.46	4,160	1,270	416	41.3	3.31	61.7	na	617	98	172	1.41	Na-Cl
dor9	Agua Caliente	Creek	356362	5939441	1,739	47.8	7.23	4,800	87	454	28.4	1.57	24.4	na	704	112	62.2	1.39	Na-Cl
dor10	Atreuco	Creek	355327	5935187	1,668	16.4	8.08	32	nd	3.1	0.7	0.56	3.1	na	0.5	15	0.94	0.02	Ca(Na)-HCO ₃
dor11	Matancilla	Creek	350995	5924804	1,301	21.2	8.47	49	nd	4.9	2.2	3.10	7.9	na	0.8	50	1.11	0.07	Ca-HCO ₃
dor12	Rio Varvarco	Rio	350368	5921396	1,184	18.0	8.04	875	12,700 [§]	97	9.1	1.72	67.1	0.78	149	56	133	0.51	Na-Cl
dos1	El Humazo	Cold spring	361866	5942377	2,198	12.9	6.80	80	nd	3.1	1.3	0.85	4.5	nd	0.4	28	1.39	0.04	Ca-HCO ₃
dos2	Los Tachos	Cold spring	361906	5938262	2,050	8.1	na	78	nd	13.7	1.2	0.34	3.9	0.10	19.1	15	3.1	0.06	Na-Cl
dot3	El Humazo	Thermal spring	361002	5942747	1,991	97.2	8.32	9,600	nd	1280	129.0	0.07	37.5	12.37	2020	78	219	3.78	Na-Cl
dot4	Los Tachos	Thermal spring	361901	5938273	2,043	96.2	na	na	nd	1160	67.5	0.60	58.7	11.16	1820	119	187	3.42	Na-Cl
dot7	Las Olletas	Thermal spring	358400	5941715	1,870	97.0	8.00	7,860	nd	1020	57.0	0.51	36.1	9.91	1590	102	160	2.98	Na-Cl
dot9	Agua Caliente	Thermal spring	356511	5939451	1,762	67.5	6.81	4,600	nd	630	40.3	1.04	26.7	5.54	945	103	96	1.85	Na-Cl

model (Lawless and Browne, 2001). Injection of a small batch of at depth, and that the progressive flashing has followed a top-down

able to infer the presence of significant amounts of fluids and energy role in triggering hydrothermal eruptions. Nevertheless, it is reasongeothermal fields (Facca and Tonani, 1971) likely play an important local sealing of the near-surface channels from where the hydrother-Domuyo volcanic complex. The 2003 explosions were attributed to Calientes and Las Olletas (Fig. 3d), characterizes the SW part of the those at Rincón de la Papas, La Bramadora, Los Tachos (Fig. 3c), Aguas This explosive event took place in a fumarolic area that, along with ton were displaced by the burst. A vapor plume with a maximum height Humazo ("The Great Smoke") (Fig. 3a,b), during which blocks up to

magma (dyke), able to destabilize the hydrothermal system, cannot

mal fluids are discharged (Mas et al., 2009). Self-sealing effects in

 $na = not \, analyzed; \, nd = not \, determined; \, EC = electrical \, conductivity.$ §This value represents the mean flow rate recorded during the measurements (see text for further explanations).

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Fig. 3. Images of thermal discharge from the SW slope of Domuyo volcano: (a and b) El Humazo; (c) Los Tachos; (d) Las Olletas.

3. Material and methods

A survey of the thermal features of the SW part of Domuyo volcano, where the 2003 hydrothermal explosions occurred, was performed in March 2013 by an Argentine-Italian team. This sector of the volcano was previously recognized (Palacio and Llambias, 1978) as the site of an important hydrothermal activity characterized by many thermal discharges, among which we selected the main ones. In particular, four thermal springs from El Humazo (dot3), Los Tachos (dot4), Las Olletas (dot7) and Agua Caliente (dot7) were sampled and analyzed for the main chemical compounds, as was Rio Varvarco upstream (dor6) and downstream (dor12) of the hot water inflow and the main creeks (from north to the south, Ahilinco: dor7; Manchana Covunco: dor1, dor2, dor8; Agua Caliente: dor9, Covunco: dor4, dor5; Atreuco: dor10; Matancilla: dor11) discharging into the Rio Varvarco. Two cold springs discharging close to El Humazo (dos1) and Los Tachos (dos2) (Fig. 2) were also sampled.

Temperature, pH and electrical conductivity were measured in the field. Water samples were filtered using 0.45 μm filters and collected in 125 and 50 mL polyethylene bottles for the determination of anions (HCO $_3$, Cl, SO $_4$) and cations (Na, K, Ca, Mg, NH $_4$ and Li). The latter aliquot was acidified with 0.5 mL of ultrapure HCl (30%). Bicarbonate was determined by acidimetric titration with 0.01 N HCl, using methylorange as indicator. Anions and cations were determined by ion chromatography using a Dionex ICS-3000 at the Laboratory of Istituto Nazionale di Geofisica e Vulcanologia (Naples, Italy). All determinations were referred to standard solutions calibrated versus NIST Standard Reference Materials. All the laboratory analytical methods and the alkalinity determinations have accuracy better than 3%.

Flow rate measurements were made in Rio Varvarco, upstream (dor6) and downstream (dor7) of the thermal inflow, and in the thermally heated creeks (dor5, dor8 and dor9). Owing to the unavailability of standard stream gaging equipment, flow rates were calculated in suitable sections using the float method. Selected sections of each river were measured in detail, and the water velocity was then computed as the average of 10 or more values obtained by measuring the travel time of a float across a fixed distance (generally, 10 to 20 m). The maximum uncertainty of the flow rate measurement of dor12 (Rio Varvarco downstream) deserves further explanations because this measurement is used to estimate the total energy release of Domuyo geothermal system. The cross section of dor12 is shown in Fig. 4 where the mean superficial velocity of the entire section (1.54 m s⁻¹), of the central fastest zone (1.89 m s $^{-1}$) and of the two lateral zones of lower velocity (1.33 m s^{-1}) is shown. In our computation, we assumed the flow rates computed with the lower and higher velocity as reasonable limits of the measurement uncertainty (11,000 and 15,700 kg s $^{-1}$). This range will be used to quantify the uncertainty of the thermal energy release

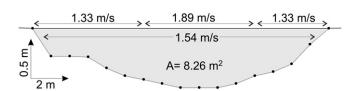


Fig. 4. The outflow area (A in m²) of the Rio Varvarco (dot12 in Table 1) and the velocities (v in m/s) used to calculate the flow rate (Q) and its uncertainties by $Q = A \times v$ (see text for explanation).

estimated by the Cl-inventory method. The basis of the Cl-inventory method is simultaneous measurement of flow rate and chemical composition of streams and rivers at specific sampling locations. Most high-chloride hot springs in non-arid regions occur near perennial streams that eventually capture most of the thermal fluid (Taran and Peiffer, 2009). The total discharge from hot-spring areas can be monitored on the basis of downstream increases in the solute loads of nearby streams, chloride being the most conservative element with which thermal waters are usually enriched relative to nearby surface and/or shallow ground waters.

4. Results

Dot3, dot4 and dot7 thermal springs discharge between 1,870 and 2,040 m of altitude and have boiling temperatures (up to 97 °C) and alkaline pH (slightly higher than 8), while dot9 has a temperature of 67.5 °C and a pH value of 6.8. Salinities (expressed as TDS, Total Dissolved Solids) are 3.78, 3.42 and 2.98 g kg-1, respectively. Agua Caliente (dot9) discharges at 1,792 m and has lower temperature (67.5 °C), pH (6.81) and salinity (1.85 g kg $^{-1}$) lower than the other thermal waters. All the thermal waters have a Na-Cl composition (Table 1). A similar composition is observed for the creeks (dor1, dor2, dor4, dor5, dor8, dor9 and dor12) into which these thermal waters discharge. Nevertheless, their salinity (from 0.06 to 1.8 g kg $^{-1}$) is lower thermal springs due to dilution processes affecting the thermal component.

Samples dor7, dor10, dor11 and dos1 are characterized by a Ca(Na)-HCO₃ composition and low salinities (\leq 0.11 g L⁻¹), while only sample dor6 has a Ca-SO₄ composition (TDS = 0.29 g L⁻¹), likely derived from the dissolution of the gypsum present upstream of the surveyed area (Palacio and Llambias, 1978).

The calculated flow rates for dor2, dor4, dor6, dor8, dor9 and dor12 are reported in Table 1. The highest value was recorded at Rio Varvarco downstream (dor12, mean value: 12,700 kg s $^{-1}$) where all the tributaries, including those that collect the thermal waters, have coalesced.

5. Discussion

5.1. Water chemistry and geothermometry

Equilibrium reservoir temperatures were calculated for the main thermal water discharges, i.e., El Humazo (dot3, $T=97\,^{\circ}$ C), Los Tachos (dot4, $T=96\,^{\circ}$ C), Las Olletas (dot7, $T=97\,^{\circ}$ C) and Agua Caliente (dot9, $T=67.5\,^{\circ}$ C) (Table 1). These Na-Cl thermal waters have chloride contents from 950 mg kg $^{-1}$ (dot9) to 2,000 mg kg $^{-1}$ (dot3). Samples dot3, dot4 and dot7 are at boiling point and are affected by shallow steam separation, which likely increases their original Cl content.

This is particularly evident at El Humazo spring where there are nearby steam vents (Fig. 3a,b). During steam separation, CO₂ is lost, causing an increase in pH (up to 8.3 in dot3) and precipitation of CaCO₃, as testified by the presence of travertine deposits. At El Humazo, travertine forms a deposit of 500×200 m with a maximum thickness of 50 m (Mas et al., 2009). Steam separation does not seem to affect Agua Caliente, the temperature of which is below the boiling point and which is characterized by the lowest chloride content (950 mg kg⁻¹) among the studied thermal waters. Agua Caliente can be considered as derived by mixing between the original thermal component and cold groundwater. Despite the possible steam separation or mixing processes, the four thermal waters are suitable for application of the Na-K-Mg geothermometer because of their very low Mg contents (from 0.07 to 1 mg kg^{-1}), which are compatible with a "mature" geothermal liquid (Giggenbach, 1988). This geothermometric method compares, in a triangular diagram, the relative proportions of the measured values (in $mg kg^{-1}$) with the theoretical ratios among Na, K and $Mg^{1/2}$ evaluated for different temperatures (Fig. 5; Giggenbach, 1988). The theoretical

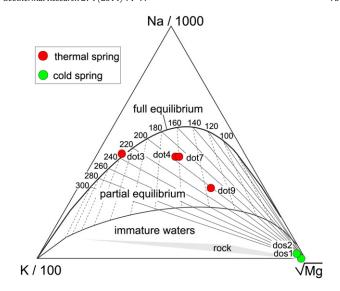


Fig. 5. The Na-K-Mg geothermometric triangular diagram (Giggenbach, 1988) and the Domuyo thermal waters. Cold waters dos1 and dos2 are reported for comparison.

values refer to a solution in equilibrium with a typical hydrothermal mineral assemblage that considers the following reactions:

$$K - feld + Na^{+} = Na - feld + K^{+}$$
(1)

These reactions refer to a mineralogical assemblage when an isochemical equilibrium is achieved. The Na/K and the K^2/Mg ratios depend on temperature according to

$$T(^{\circ}C) = \{1390/[1.75 + log(Na/K)]\}-273.15$$
 (3)

$$T(^{\circ}C) = \{4410/[14.0 - \log (K2/Mg)]\} - 273.15$$
 (4)

The intersection of each Na-K and K-Mg isotherm corresponds to water compositions in equilibrium with a mineralogical assemblage that controls both geothermometers and delineates the so-called "full equilibrium" curve. In Fig. 5, the fields of partial equilibrium and immature waters are also delineated. Because of the fast kinetics of the reactions involving Mg, Eq. (4) can quickly equilibrate at the decreasing temperatures encountered by a geothermal liquid ascending toward the surface, displacing the points in Fig. 5 toward the Mg corner. The effect is similar to that caused by mixing of the thermal component with shallow cold groundwater which, relative to the geothermal liquids, is enriched in Mg. However, such secondary processes do not substantially change the temperature estimates based on the Na-K geothermometer, the latter providing reliable estimation of the temperature of the original thermal liquid. The Domuyo thermal waters lie either along the "full equilibrium" curve (dot3) or in the "partial equilibrium" field (dot4, dot7, dot9), indicating Na/K temperatures of 233 °C (dot3), 192 °C (dot4), 190 °C (dot7) and 199 °C (dot9). As suggested by the relatively low Cl content and emergence temperature, dot9 (Agua Caliente) is likely sourced by a thermal system that mixes with shallow cold groundwater of composition similar to samples dos1 and dos2 (Fig. 5). The other thermal springs, which emerge at the boiling point, plot close to the full equilibrium curve. In these cases, the significant

secondary process is loss of steam during boiling, which does not affect the Na/K ratio, but increases the absolute concentrations of ions.

5.2. Flux of the geothermal liquid in the Rio Varvarco and its tributaries and computation of the total thermal release

The relatively alkaline pH values and high Cl concentrations suggest that the Domuyo thermal waters may be affected by steam loss during boiling. Such a process is particularly evident for dot3 (El Humazo) where vigorous vapor vents accompany the water discharge (Fig. 3a, b). According to Mas et al. (2009), the emitted gas consists of 99% of steam and 1% of other gases, among which CO₂ is the most abundant (JICA, 1983).

Assuming an iso-enthalpic boiling process, the fractions of separated steam (from 0.17 to 0.26) were subsequently used to restore the original composition of the un-boiled liquid by referring to the mobile species (Cl ~ 1,500 mg kg $^{-1}$, Na ~ 950 mg kg $^{-1}$, Li ~ 9.2 mg kg $^{-1}$ for dot3 and dot4, and Cl ~1,300 mg kg $^{-1}$, Na ~ 850 mg kg $^{-1}$, Li ~ 8.2 mg kg $^{-1}$ for dot2). The restored compositions of the three samples are similar and in particular those of sample dot3 and dot4 are practically the same. The restored concentration of Cl from dot3 and dot4, which are characterized by the highest flow rate, is our reference concentration for the un-boiled liquid and is assumed as the minimum concentration in computation of the thermal energy. The measured Cl concentration of dot3 (2,000 mg kg $^{-1}$) was considered as the maximum concentration of the original geothermal liquid, assuming a non-isenthalpic process where the cooling of the water is governed by heat exchange with the rocks, without any steam separation.

The conservative behavior of Na, Cl and Li may allow use of these solutes as tracers of the thermal component flux in Rio Varvarco and its tributaries. We assumed that Cl, Na and Li contents increase from Rio Varvarco upstream (dor6) to Rio Varvarco downstream (dor12) is due to the hot waters in the tributaries, i.e., Manchana Covunco (dor8: 29.7 °C, 4,160 μ S/cm), Agua Caliente (dor9: 47.8 °C, 4,800 μ S/cm) and Covunco (dor5: 23.1 °C, 2,510 μ S/cm). The other tributaries have ambient temperature and lower salinity as suggested by their low electrical conductivity (dor7: Ahilinco 11.5 °C, 147 μ S/cm; dor10, Atreuco 16.4 °C, 32 μ S/cm; dor11, Matansilla: 21.1 °C, 49 μ S/cm).

Qualitatively the mixing between the thermal component and the low salinity waters is indicated by the very high positive correlations ($R^2 = 0.999$) among Na, Cl and Li, as shown in Fig. 6a,b. In these figures, the computed Cl concentration of the pre-boiled geothermal liquid is also shown.

The fluxes of Na, Cl and Li were computed by multiplying their concentrations by the respective flow rates measured at Rio Varvarco (upstream and downstream) and Manchana Covunco, Agua Caliente and Covunco hot creeks. For each species, the sum of the creek fluxes (Na: 960 g s⁻¹; Cl: 1,450 g s⁻¹; Li: $8.5 g s^{-1}$) is systematically lower than those computed in the Rio Varvarco as the difference between the downstream and upstream values (Na: 1,190 g s $^{-1}$; Cl: 1,880 g s $^{-1}$; Li: 9.9 g s^{-1}). This systematic difference, though less than 25%, could be due to some uncertainties in the flow rate computation. Alternatively, part of the creek water may be lost as infiltration into the permeable alluvial deposits but later seeps into the river downstream from the confluence or some groundwater inflow to the Rio that includes a thermal component. Nevertheless, we think that the observed differences are acceptable. Assuming that the flux calculated for Rio Varvarco is the most reliable, the computation of the thermal energy release (QH) and related uncertainties were performed using the simple relation:

$$Q_{H} = (Q_{CI}/CI_{t}) \times H_{t} \tag{5}$$

where Q_{CI} is the flux of CI in the river (g s⁻¹), CI_t is the CI concentration in the original thermal liquid (g g⁻¹, the unity has been converted from mg kg⁻¹) and H_t is the enthalpy (J g⁻¹) of the liquid at the estimated reservoir temperature.

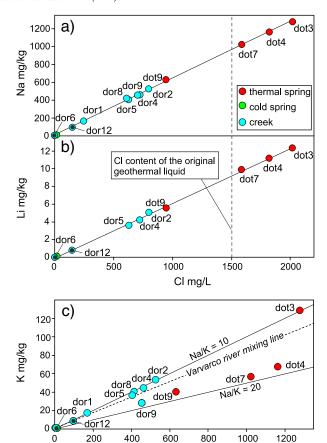


Fig. 6. Binary diagrams (a) Na vs. Cl, (b) Li vs. Cl and (c) Na vs. K. Symbols as in Fig. 2. The unlabelled samples are diluted waters close to the axis origin in each diagram.

Na mg/kgL

We solved Eq. (5) with a Monte Carlo procedure considering the uncertainties of the flow rate of Rio Varvarco (11,000 and 15,700 kg s $^{-1}$) and of Cl_t (1.5–2.0 g kg $^{-1}$). The enthalpy H_t was computed for an original temperature of 220 °C \pm 10 °C (enthalpy = 944 J g $^{-1}$ \pm 46 J g $^{-1}$), i.e., the temperature resulting from the average Na/K ratio of the thermal tributaries which enter Rio Varvarco as reported in the Na-K diagram of Fig. 6c. The Monte Carlo simulation gives a thermal water total discharge of 1,150 \pm 150 kg s $^{-1}$ and a thermal energy release of 1.1 \times 109 \pm 0.2 \times 109 W (1.1 \pm 0.2 GW).

6. Conclusion

A geochemical survey of the main thermal waters discharging in the southwestern part of the Domuyo volcanic complex has highlighted the strikingly high thermal energy release from this remote site in Patagonia. A simple hydrogeochemical approach shows that ~1,150 kg s $^{-1}$ of boiling water is conveyed to the main drainage system (Rio Varvarco) through its tributaries. Referring to a liquid originally at 220 °C, the calculated thermal energy release is as large as ~ 1.1 GW, a value that, to the best of our knowledge, is the second largest advective heat flux ever measured from a single onland volcanic center (other than Yellowstone). This result suggests an enormous geothermal potential of the Domuyo system, which has yet to be exploited.

A previous geochemical survey of the same thermal waters, performed in 1976 (Llambías et al., 1978), reports chemical compositions, temperatures and thermal features similar to those measured in our work, suggesting a persistent heat flux. In addition to local sealing of hydrothermal channels, this large flux of energy may be the main reason

for the 2003 hydrothermal explosions at El Humazo. Finally, this very large heat flux is somewhat hard to explain in terms of either the cooling of mapped magmatic intrusions, which seem to have occurred at 0.72 Ma, or the most recent volcanic activity at 0.11 Ma. In our opinion, the presence of an undocumented, younger volcanic activity cannot be excluded. Alternatively, the high heat flux may originate from a magma intrusion at shallow depth, possibly related to a recent reactivation of the Domuyo volcanic system.

Acknowledgements

The authors are gratefully to the personnel of the Domuyo Provincial Park and to our guide Paulo Fanti. Many thanks are due to the four Brazilian students (Vanessa Araujo, Cristiane Schmidt, Marcello Cappelli and Hugo Vidal) who accompanied and helped us during the fieldwork. The authors thank William C. Evans and an anonymous reviewer for their useful comments and suggestions on the early version of the manuscript.

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