RESEARCH ARTICLE



The 2012–2016 eruptive cycle at Copahue volcano (Argentina) versus the peripheral gas manifestations: hints from the chemical and isotopic features of fumarolic fluids

F. Tassi^{1,2} • M. Agusto³ • C. Lamberti³ • A. T. Caselli⁴ • G. Pecoraino⁵ • C. Caponi¹ • J. Szentiványi³ • S. Venturi^{1,2} • O. Vaselli^{1,2}

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Abstract This study presents the chemical and isotopic compositions of hydrothermal gases from fumaroles discharging around Copahue volcano (Argentina). Gas samples, including those from two fumaroles at the active summit crater, were collected during 13 surveys carried out by different research teams from 1976 to February 2016. The time-series of H₂, CO and light hydrocarbons showed episodic increases related to the main events of the last eruptive cycle that started on 19 July 2012. Concentration peaks were likely caused by enhanced input of hot magmatic fluids affecting the hydrothermal reservoir. These data contrast with the temporal variations shown by Rc/Ra and δ^{13} C-CO₂ values in 2012–2014, which indicated an increasing input from a crustal fluid source. In 2015–2016, however, these isotopic parameters showed opposite trends; their composition became closer to that of the two summit fumaroles, which possibly corresponds to that of the deep magmatic-related end-member. The delayed and reduced compositional changes in the peripheral hydrothermal fluid

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F. Tassi franco.tassi@unifi.it

- ¹ Department of Earth Sciences, University of Florence, Via La Pira, 4, 50121 Florence, Italy
- ² Institute of Geosciences and Earth Resources of the National Research Council (CNR-IGG), Via La Pira, 4, 50121 Florence, Italy
- ³ GESVA-IDEAN, Dpto. Cs. Geológicas, FCEN, Universidad de Buenos Aires, Ciudad Universitaria, Pab.2, 1428 Buenos Aires, Argentina
- ⁴ LESVA, Universidad Nacional de Río Negro, Roca 1242, 8332 General Roca, Argentina
- ⁵ Istituto Nazionale di Geofisica e Vulcanologia, Via U. La Malfa 153, 90146 Palermo, Italy

discharge in response to the 2012–2016 eruptive events suggest that geochemical surveys of these emissions are unlikely to provide premonitory signals of volcanic unrest if the volcanic activity remains centered in the main crater. Instead, an instrument which is able to provide measurements of volcanic gases in the air (e.g. MultiGAS) may be used to detect changes at the summit crater. Otherwise, monitoring of seismic activity and ground deformation, as well as the periodic measurement of the chemistry of the water in the Rio Agrio, which is fed by thermal discharge from the summit crater, seem to represent the most reliable means of monitoring at Copahue. However, the relative compositional stability of the hydrothermal reservoir is a great advantage in terms of geothermal resource exploitation and could encourage new investments in the Copahue geothermal project which was abandoned in the 1990s.

Keywords Geochemical monitoring · Copahue volcano · Fluid geochemistry · Hydrothermal system · Active volcano

Introduction

Fumarolic gases discharging from hydrothermal reservoirs hosted in quiescent volcanoes carry sensitive compounds (e.g. H_2 , CO, CO₂, CH₄ and H_2O) that provide the means for investigating the chemical-physical conditions of the deep environment (e.g. Giggenbach 1987, 1996; Chiodini and Marini 1998). Noble gas isotopic signatures are useful tracers for identifying fluid source regions, especially in subduction zones (e.g. Poreda and Craig 1989; Giggenbach et al. 1993; Fischer et al. 2002; Hilton et al. 2002, 2010; Shaw et al. 2003; Sano and Fischer 2013). Accordingly, fluid discharge is periodically monitored to assess the degassing behavior of magmatic bodies (e.g. Menyailov et al. 1986; Tassi et al. 2016) and changes in volcanic activity (e.g. Chiodini et al. 1995; Fischer et al. 1996, 2015; Symonds et al. 1996; Vaselli et al. 2010; Melián et al. 2012). On the other hand, innovative remote sensing techniques, able to continuously acquire chemical data of gases, provide valuable insights into the behavior of active volcanoes and eruption forecasting (e.g. Aiuppa et al. 2009). Although the monitoring of magmatichydrothermal fluids has for many decades been considered a robust prognostic tool for detecting rejuvenation of volcanic activity (e.g. Stoibert and Rose 1970), it is not surprising that long-term geochemical time-series datasets are available for only a relatively small number of volcanic systems. This is particularly true for volcanoes located in developing countries, where logistic and economic problems often hinder the development of long-term research projects, which are frequently operated and funded by western countries.

The geochemistry of fluids discharged from geothermal wells and natural emissions located in the surroundings of the main cone of Copahue volcano (Argentina) has been investigated since the 1970s within the framework of geothermal exploration projects carried out by private companies (Mas and Mas 2016 and references therein). Geochemical studies were performed on this system from 2006 to 2012, with the aim of constructing a comprehensive conceptual geochemical model describing the primary fluid sources and the secondary processes controlling the chemistry of the fluid discharge (Agusto et al. 2013; Tassi et al. 2016). A new eruptive cycle, which began on 19 July 2012 (after 12 years of quiescence), was characterized by occasional phreatic and phreatomagmatic eruptions (Caselli et al. 2016a); it promoted further surveys, providing the opportunity to upgrade and improve the geochemical database of fluids emitted from this volcano.

In this paper, we present both new and previously published chemical and isotopic (${}^{3}\text{He}/{}^{4}\text{He}$ and $\delta^{13}\text{C}$ in CO₂) data for gases from the main fluid discharge sources of Copahue volcano. This unique geochemical dataset covers, albeit discontinuously, a 40-year observation period (from 1976 to 2016) and is used to describe the relationships between the ongoing volcanic activity and the temporal evolution of the hydrothermal reservoir feeding the surficial emissions. In addition, the chemistry of two fumaroles sampled in March 2013, which are currently to the best of our knowledge the only available gas samples from the volcano's summit crater, is also reported.

Geological and volcanological settings

Volcanism in the Southern Andean Volcanic Zone (SAVZ) is related to the subduction of the Nazca Plate beneath the South American Plate (Pesicek et al. 2012 and references therein). Copahue volcano belongs to the Caviahue-Copahue Volcanic Complex (CCVC, 38°S-71°W; Argentina-Chile) that is located at the transition between the southern strike-slip Liquiñe-Ofqui and the northern compressive Copahue-Antiñir Fault systems (Fig. 1a) (Cembrano et al. 1996, 2002; Stern 2004; Radic 2010). Early Pliocene basaltic-andesitic eruptions associated with extensional and/or transtensional processes ended in the latest Pliocene-early Pleistocene, with the formation of a series of calderas which discharged large amounts of ignimbrite (Folguera et al. 2016). The construction of the CCVC volcanic edifice, which has an elongated NE-SW shape and lies on basement consisting of volcanic sequences belonging to the Cola de Zorro Formation, began ~ 1.23 Ma in the western sector of one of the Caviahue Caldera. This relatively young stratovolcano was intruded by the 1 Ma dacitic Pucón Mahuida dome, whose volcanic products were covered by pre-glacial lavas, ignimbrites and tuffs (1-0.7 Ma; Copahue Stage 1) and syn- and post-glacial pillow lavas (700-15 ka; Copahue Stage 2). Volcanic products of Copahue Stage 3 (<~ 15 ka) consist of lavas erupted from post-glacial vents located along the northern slope and the highest, most easterly crater of the stratocone (Melnick et al. 2006).

The 200 m-wide easternmost crater of the Copahue summit, which has nine N60°E-aligned craters, has experienced 13 phreatic and low-magnitude phreatomagmatic eruptive episodes over the last 260 years in 1750, 1759, 1867, 1937, 1944, 1960, 1961, 1992, 1993, 1994, 1995, 2000 and 2012– present (Caselli et al. 2016b and references therein).

The 2012–2016 eruptive cycle

Precursory signals for the 2012–2016 eruptive cycle mainly consisted of enhanced high- and low-frequency seismic activities, which followed the Mw 8.8 Maule earthquake (February 27, 2010) (Forte et al. 2012). At the end of 2011, a 200-300 m-high column of dense vapor was produced by the summit crater, followed by a progressive increase in seismic signals (INPRES, Araujo, personal communication). Significant changes in temperature, pH and chemical composition of the hyperacidic lake located within the summit crater were also recorded. On 17 July 2012, a small-scale phreatic explosion from the crater lake was observed, followed two days later by phreatomagmatic eruptions that produced a small E-SE-oriented plume of about 18 km in length (Caselli et al. 2016a). In August-November 2012, the fumarolic activity in the summit crater was persistently intense, causing a decrease in the lake water level of ~ 20 m. On 22 December 2012 at 12:05 (GMT), an 800 m-high white vapor column was emitted from the active crater that, in a few hours, caused the complete evaporation of the crater lake. Ash columns and incandescent material which was distributed over an area of $\sim 8700 \text{ km}^2$ to the SE were produced during the following phreatomagmatic and strombolian eruptions

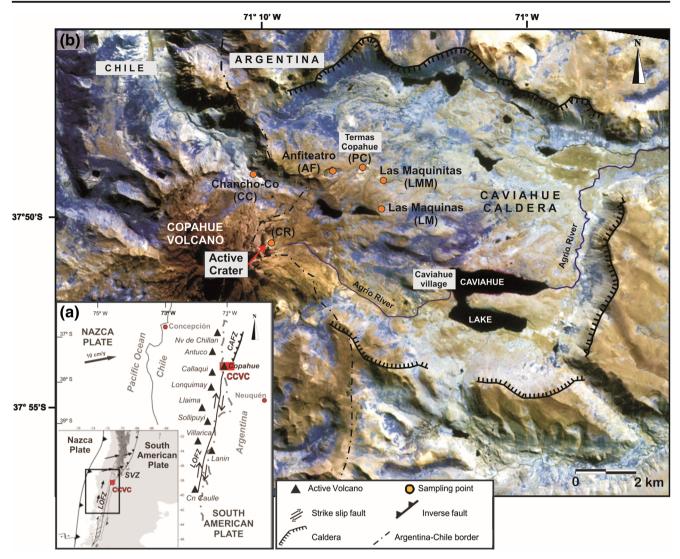


Fig. 1 (a) Schematic map of the northern transitional and central segments of the Southern Andean Volcanic Zone with location of the main active volcanoes and Caviahue-Copahue Volcanic Complex

(CCVC); (**b**) map of CCVC area with location of the gas sampling sites (modified from Agusto et al. 2013)

(Petrinovic et al. 2014; Caselli et al. 2016a). In 2013, no significant volcanic events were observed. This pause allowed, by the end of 2013, the refilling of the crater lake, although its volume was significantly smaller than that recorded before the summer of 2012. This period of quiescence lasted until July 2014, when a phreatic explosion generated lahars on the eastern flank of the volcanic edifice. In early October 2014, the lake once again dried out due to the occurrence of a phreatomagmatic eruption with the emission of ash and volcanic bombs, containing sulfur particles and quenched liquid sulfur, respectively (Caselli et al. 2016a and references therein). In January-February 2015, the presence of a new crater lake was observed. In October 2015, a strong phreatomagmatic to magmatic eruption occurred. After this event, a small new crater lake, which was still present during the sampling fieldtrip carried out in February 2016, formed again.

Conceptual model of the Copahue hydrothermal-magmatic system

Varekamp et al. (2004, 2006) suggested that a magmadominated brine was feeding the crater lake, the latter receiving fluids from a magma batch at ~ 4–5 km depth (Velez et al. 2011). The strong relationship between (a) the lake's geochemical features and volume and (b) the rate of magmatic gas inputs was recently confirmed by Agusto and Varekamp (2016), who based their assessment on 15 years (1997–2012) of periodical observations. The release of magmatic fluids from the Copahue crater even during periods of quiescence was confirmed in March 2013 by both the analysis of two crater fumaroles, included in the present paper, and a variety of in situ and remote sensing measurements carried out in March 2014, which identified a plume released from the crater lake which was discharging up to ~ 1400 t/day of SO₂ (Tamburello et al. 2015).

There are five distinct areas around the volcanic edifice. where thermal fluid emissions, consisting of fumaroles and boiling-bubbling water and mud pools occur: Las Máquinas (LM), Las Maquinitas (LMM), Termas de Copahue (PC), Anfiteatro (AF) and Chancho Co (CC) (Fig. 1b). The origin of the fluids discharged from these peripheral emissions was firstly investigated within the framework of geological, geochemical and geophysical surveys carried out since 1974 for a preliminary evaluation of the area's geothermal potential (Jurío 1977; Panarello et al. 1988; Panarello 2002; JICA 1992). The chemical and physical characteristics of fluids from four wells drilled at different times (1976, 1981 and 1998) ~ 5 km NW of the volcano summit (Mas and Mas 2016 and references therein) indicated the occurrence of two main hydrothermal aquifers located at 800-1000 and > 1400 m depth. The latter had a temperature of up to 250 °C, as also confirmed by geothermometric estimations in the H₂O-CO₂-CH₄-C₃H₆-C₃H₈ system (Tassi et al. 2016). This multi-layer peripheral reservoir was found to have been produced by the interaction between magmatic gases and meteoric water (Agusto et al. 2013). This process is responsible for (a) the dissolution of strongly acidic gases from magma degassing (SO₂, HCl and HF) and (b) the production of the typical hydrothermal gases, i.e. H₂, H₂S, CO, CH₄ and light hydrocarbons.

Gas sampling and analytical methods

Gas samples were collected from selected fumaroles (LM, LMM, PC, AF and CC) in the five main hydrothermal areas (Fig. 1b). No fumarolic emissions had been seen at the summit crater over the last few decades, due to the presence of the crater's lake. However, in March 2013, the lake almost completely dried out due to the high discharge rate of magmatic fluids, allowing access to the crater bottom, where two fumaroles (CR1 and CR2) were sampled (Tamburello et al. 2015).

The sampling and analytical procedures adopted for the gases constituting the new dataset (from December 2012 to March 2016) were similar to those used to produce most of the previous data (from November 2006 to March 2012; Agusto et al. 2013). The gas sampling line consisted of a 1 m-long titanium tube ($\emptyset = 2.5$ cm) and Pyrex glass-dewared pipes connected to pre-weighted and pre-evacuated 50-mL glass Thorion®-tapped flasks filled with 20 mL of a 4 N NaOH and 0.15 M Cd(OH)₂ suspension (Montegrossi et al. 2001). This technique allows SO₂ to be separated in situ and then dissolved in the alkaline solution with water vapor and acidic gases (CO₂, HCl, HF). The H₂S forms insoluble CdS, whereas low-solubility gas species (N₂, O₂, CO, H₂, He, Ar, CH₄ and light hydrocarbons) are trapped in the flask headspace. Steam condensates (for the analysis of the δ^{18} O and δ D ratios in

water vapor) and dry gases (for the analysis of the ${}^{13}C/{}^{12}C$ ratios in CO₂ and He isotopes) were collected using a watercooled condenser connected to the soda flask sampling line.

Inorganic (N₂, Ar, O₂, H₂, He and CO) and organic (C₁–C₃ hydrocarbons) gases from the flask headspace were analyzed at the Laboratory of Fluid Geochemistry of the University of Florence (Italy) by gas chromatography (GC Shimadzu 15A, Shimadzu 14 and Thermo Focus). The liquid and the solid phases of the suspension were separated by centrifuge, to analyze for CO₂, SO₂, HCl and HF, in the form of CO_3^{2-} (by acidimetric titration, AT; Metrohm Basic Titrino), SO_4^{2-} , Cl⁻ and F⁻ (by ion chromatography, IC; Metrohm 761), respectively. The CdS precipitate was dissolved by oxidation with H₂O₂ in order to analyze H₂S as SO_4^{2-} by IC. The analytical error for GC, AT and IC analyses was < 5%.

The analysis of ${}^{13}C/{}^{12}C$ of CO₂ (expressed as $\delta^{13}C$ -CO₂ ‰ vs. V-PDB) was carried out at the CNR-IGG laboratory of Stable Isotopes (Pisa, Italy) using a Finnigan MAT252 mass spectrometer after standard extraction and purification procedures of the gas mixtures (Evans et al. 1998) performed at the Department of Earth Sciences of Florence (Italy). Carrara and San Vincenzo marbles, as well as international NBS18 and NBS19 standards, were used to estimate the external precision. The analytical error and the reproducibility were \pm 0.05 and \pm 0.1‰, respectively.

The *R/Ra* (where *R* is the ³He/⁴He measured ratio and *Ra* is the ³He/⁴He ratio in the air 1.39×10^{-6} ; Mamyrin and Tolstikhin 1984) and ⁴He/²⁰Ne ratios were determined at the INGV laboratories in Palermo (Italy) by separately introducing He and Ne into a split-flighttube mass spectrometer (GVI Helix SFT) after performing standard purification procedures (e.g. Rizzo et al. 2015). The analytical error was $\pm 1\%$. The measured *R/Ra* values were corrected for air contamination using the ⁴He/²⁰Ne ratios (Poreda and Craig 1989), as follows:

$$Rc/Ra = [(R/Ra)-r)]/(1-r)$$
 (1)

where $r = ({}^{4}\text{He}/{}^{20}\text{Ne})_{air}/({}^{4}\text{He}/{}^{20}\text{Ne})_{meas}$, the $({}^{4}\text{He}/{}^{20}\text{Ne})_{air}$ ratio being that in the atmosphere (0.318; Ozima and Posodek 1983) and the $({}^{4}\text{He}/{}^{20}\text{Ne})_{meas}$ ratio that measured in the gas sample.

The ¹⁸O/¹⁶O and ²H/¹H ratios of water vapor (expressed as δ^{18} O-H₂O and δ D-H₂O %_o vs. V-SMOW, respectively) were analyzed at the Department of Physics and Earth Sciences of the University of Parma (Italy) using a Thermo-Finnigan HDO equilibrator coupled to a Thermo-Finnigan Delta Plus mass spectrometer with double entry system. The analytical errors were \pm 0.05 and \pm 1%_o for δ^{18} O-H₂O and δ D-H₂O, respectively.

Results

Chemical composition

Outlet temperatures, chemical composition (dry fraction) and steam concentrations (in mmol/mol for the main species, in μ mol/mol for all hydrocarbons except CH₄) of the fumarolic gases collected from December 2012 to March 2016 are reported in Table 1. Data from the literature for gases collected from the emissions in 1976 (Jurío 1977), 1986 (Panarello 2002), 1997 (Martini et al. 1997), November 2006–March 2012 (Agusto et al. 2013) and January 2014 (Tardani et al. 2016) are also listed in Table 1.

The LM, PC, AF and CC fumaroles were characterized by outlet temperatures \leq 97 °C, whereas those of LMM ranged from 130 °C (in 1986) to 160 °C (in March 2012). The CR1 and CR2 outlet temperatures were 431 and 102 °C, respectively. The chemical composition of the five distal discharges was typical of hydrothermal fluids (e.g. Giggenbach 1987; Chiodini and Marini 1998), mainly consisting of water vapor $(\geq 960 \text{ mmol/mol})$ with a dry gas fraction dominated by CO₂ (up to 974 mmol/mol) and variable concentrations of H₂S (2.5 to 155 mmol/mol), N_2 (3 to 169 mmol/mol), H_2 (1 to 105 mmol/mol) and CH₄ (0.68 to 55 mmol/mol). He, Ar and CO were present in minor amounts (up to 0.089, 0.31 and 0.018 mmol/mol, respectively). The C₂-C₃ hydrocarbons, whose total content was up to 52 µmol/mol, mostly consisted of ethane (C₂H₆, up to 47 µmol/mol) and propane (C₃H₈, up to 5.1 µmol/mol), with minor concentrations of propene (C₃H₆, up to 0.0081 µmol/mol).

The chemical features of the CR1 and CR2 fumaroles were very different compared to those of the hydrothermal-type emissions, since the former were characterized by (a) significant concentrations of magmatic-related acidic gases, such as SO₂ (165 and 120 mmol/mol, respectively), HCl (5.1 and 3.8 mmol/mol, respectively) and HF (2.9 and 2.6 mmol/mol, respectively); (b) relatively low concentration of CH₄, C₂H₆ and C₃H₈ (< 0.5 mmol/mol and \leq 0.66 and \leq 0.012 µmol/mol, respectively) and (c) presence of ethene (C₂H₄; 0.13 and 0.11 µmol/mol, respectively).

δ^{13} C-CO₂ and *Rc/Ra* values

The δ^{13} C-CO₂, *Rc/Ra*, ⁴He/²⁰Ne, δ^{18} O-H₂O and δ D-H₂O values of the fumarolic gases collected from December 2012 to March 2016 are reported in Table 2. In addition to the literature data from the authors cited above, isotopic data for one gas sample collected in 2003 (Varekamp et al. 2006) are also included.

The δ^{13} C-CO₂ values ranged from – 9.6 to – 6.7‰ vs. V-PDB, with no significant differences between the different gas emissions. On the contrary, the *Rc/Ra* values of the LM, LMM, PC and CR1 gases were higher (5.89 to 7.96) with

respect to those of the AF and CC gases (4.04 to 5.35). The high ${}^{4}\text{He}/{}^{20}\text{Ne}$ ratios (23 to 2445) indicate that the effect of air contamination on the *Rc/Ra* values was negligible. The $\delta^{13}\text{C-CO}_2$ value of the CR1 fumarole (- 9.6% vs. V-PDB) was at the lowest limit of the range shown by the hydrothermal discharges (- 6.9 to 9.6% vs. V-PDB), whereas its *Rc/Ra* value (7.75) was one of the highest measured at Copahue volcano.

The values of δ^{18} O-H₂O and δ D-H₂O in the condensate samples of LM, LMM, CC, AF and PC ranged from – 11.8 to – 8.2% and from – 92 to – 75% vs. V-SMOW, respectively, with the only exception being the AF gas sample collected in March 2012 that had more negative values (– 17.9 and – 109% vs. V-SMOW, respectively). On the contrary, the CR1 and CR2 fumaroles showed less negative values (– 3.5 and – 66% and – 3.6 and – 70% vs. V-SMOW, respectively).

Discussion

Temporal evolution of the chemical composition of the gas

Hydrothermal systems typically occur around active volcanoes, since they are produced by the interaction of hot fluids released from the magmatic bodies and shallow aquifers. This process implies that the most reactive magmatic volatiles (SO₂, HCl and HF) dissolve when entering the liquid reservoir. Thus, they are commonly not identified in the hydrothermal fluid discharges. However, increasing inputs of heat and magmatic gases are expected to cause a strong perturbation in the hydrothermal-magmatic system, allowing the most soluble species, although partially scrubbed, to be released from the fumarolic discharge. Accordingly, geochemical monitoring of fumarolic fluids, including those from peripheral hydrothermal-type emissions, may record precursory signals related to physical-chemical changes occurring at depth prior to a new phase of volcanic activity. This was the case at Turrialba volcano (Costa Rica), where SO₂ was firstly detected in low temperature fumaroles of the summit craters in September 2001, after three consecutive years of observations. This event was interpreted as being precursory to a volcanic crisis (Tassi et al. 2004), which actually happened almost 10 years later (Vaselli et al. 2010) and is still ongoing (de Moor et al. 2016a). Similarly, in the four months following the impressive gas burst that occurred on 3 November 2002 (Capaccioni et al. 2005), the submerged fumaroles located next to Panarea Island, a dormant volcano in the Aeolian Archipelago (southern Italy), discharged SO₂-bearing gases that had never previously been detected in these fumaroles (Capaccioni et al. 2007 and references therein), revealing the temporary consumption of the liquid-dominated aquifer feeding the submarine fumarolic field likely due to a magmatic gas pulse from depth. On the other hand, experimental

Table 1Outlet temperature (in °C), chemical composition of the cspecies are in mmol/mol, whereas those of the hydrocarbons exceptineet al. (2013) and (5) Tardani et al. (2016)	temperature ol/mol, when (5) Tardani e	Outlet temperature (in °C), chemical composition of the dry gas fraction and steam concentrations of the CCVC gases; n.d.: not detected; n.a.: not analyzed. The concentrations of the main gas in mmol/mol, whereas those of the hydrocarbons excepting CH_4 are in µmol/mol. References for the chemical data are: (1) Jurío (1977), (2) Panarello (2002), (3) Martini et al. (1997), (4) Agusto 3) and (5) Tardani et al. (2016)	nical compo he hydroca	ssition rbons e	of the d xcepting	ry gas g CH4	fraction are in μ	mol/mo	am conc l. Refere	entration nces for	is of the the the chem	CCVC g	ases; n.d are: (1) .	.: not d Jurío (1	etected; n 977), (2)	La.: not ar Panarello	lalyzed. (2002),	The conc (3) Marti	dry gas fraction and steam concentrations of the CCVC gases; n.d.: not detected; n.a.: not analyzed. The concentrations of the main gas ng CH4 are in μmol/mol. References for the chemical data are: (1) Jurío (1977), (2) Panarello (2002), (3) Martini et al. (1997), (4) Agusto	of the m 97), (4)	ain gas Agusto
Sample	Acronym	Reference	Date	T°C	CO_2	HF	HCI	SO_2	H_2S	N_2	CH4	Ar	02	H_2	He	CO	C_2H_4	C_2H_6	C_3H_6	C_3H_8	Steam
Las Máquinas	LM	1	1976	n.d.	834	n.d.	n.d.	n.d.	4.5	105	32	0.080	n.a.	24	0.013	0.0042	n.a.	n.a.	n.a.	n.a.	n.d.
Las Máquinas	LM	2	1986	85	927	n.d.	n.d.	n.d.	6.2	28	22	0.11	n.a.	14	0.013	0.0014	n.a.	n.a.	n.a.	n.a.	n.d.
Las Máquinas	LM	3	1997	n.d.	965	n.d.	n.d.	n.d.	6.1	24	1.1	0.006	n.a.	1.0	0.002	0.0001	n.a.	n.a.	n.a.	n.a.	n.d.
Las Máquinas	LM	4	Nov-06	95	971	n.d.	n.d.	n.d.	11	6.7	5.8	0.028	0.19	5.5	0.028	n.a.	n.d.	6.5	0.0005	0.56	970
Las Máquinas	LM	4	Feb-07	94	989	n.d.	n.d.	n.d.	2.5	2.7	3.7	0.010	0.025	2.5	0.015	n.a.	n.d.	2.9	0.0004	0.30	n.d.
Las Máquinas	LM	4	Mar-12	96	938	n.d.	n.d.	n.d.	8.1	17	20	0.023	0.009	16	0.015	0.0021	n.d.	23	0.0016	2.3	979
Las Máquinas	LM		Dec-12	94	948	n.d.	n.d.	n.d.	7.6	14	11	0.022	0.008	17	0.012	0.0022	n.d.	15	0.0011	1.5	975
Las Máquinas	LM		Jan-13	95	947	n.d.	n.d.	n.d.	14	16	29	0.019	0.011	35	0.011	0.0036	n.d.	13	0.0078	1.9	981
Las Máquinas	LM		Mar-13	94	953	n.d.	n.d.	n.d.	4.9	11	26	0.026	0.015	4.3	0.012	0.0018	n.d.	18	0.0023	2.8	971
Las Máquinas	LM	5	Jan-14	93	940	n.d.	n.d.	n.d.	4.4	6.7	28	n.d.	0.3	n.a.	0.011	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.
Las Máquinas	LM		Mar-14	94	953	n.d.	n.d.	n.d.	8.3	17	13	0.021	0.011	8.8	0.012	0.0031	n.d.	24	0.0018	2.1	975
Las Máquinas	LM		Mar-15	96	937	n.d.	n.d.	n.d.	12	19	16	0.018	0.009	16	0.015	0.0039	n.d.	28	0.0021	3.2	971
Las Máquinas	LM		Feb-16	96	928	n.d.	n.d.	n.d.	11	18	11	0.017	0.011	32	0.014	0.013	n.d.	25	0.0057	1.3	978
Las Maquinitas	LMM	1	1976	n.d.	905	n.d.	n.d.	n.d.	16	55	15	0.070	n.a.	16	0.014	0.0020	n.a.	n.a.	n.a.	n.a.	n.d.
Las Maquinitas	LMM	2	1986	130	944	n.d.	n.d.	n.d.	14	24	13	0.090	n.a.	14	0.010	0.0013	n.a.	n.a.	n.a.	n.a.	n.d.
Las Maquinitas	LMM	3	1997	n.d.	936	n.d.	n.d.	n.d.	3.0	26	21	0.042	n.a.	Э	0.003	0.0004	n.a.	n.a.	n.a.	n.a.	n.d.
Las Maquinitas	LMM	4	Nov-06	135	974	n.d.	n.d.	n.d.	7.8	8.7	4.0	0.026	0.087	6.2	0.029	n.a.	n.d.	6.1	0.0006	0.67	962
Las Maquinitas	LMM	4	Mar-12	160	944	n.d.	n.d.	n.d.	8.6	21	13	0.021	0.006	13	0.019	0.0021	n.d.	11	0.0009	1.1	971
Las Maquinitas	LMM		Dec-12	155	933	n.d.	n.d.	n.d.	9.2	25	16	0.033	0.008	17	0.012	0.0023	n.d.	12	0.0011	1.0	970
Las Maquinitas	LMM		Jan-13	148	910	n.d.	n.d.	n.d.	15	27	26	0.038	0.007	28	0.013	0.0039	n.d.	13	0.0081	1.2	981
Las Maquinitas	LMM		Mar-13	131	920	n.d.	n.d.	n.d.	9.5	29	22	0.037	0.011	19	0.014	0.0026	n.d.	11	0.0012	1.1	981
Las Maquinitas	LMM		Mar-14	133	945	n.d.	n.d.	n.d.	7.3	23	14	0.025	0.008	11	0.011	0.0015	n.d.	14	0.0013	1.3	975
Las Maquinitas	LMM		Mar-15	136	924	n.d.	n.d.	n.d.	8.7	25	16	0.031	0.016	26	0.013	0.0026	n.d.	17	0.0015	2.1	983
Las Maquinitas	LMM		Feb-16	135	906	n.d.	n.d.	n.d.	9.3	25	12	0.023	0.009	48	0.015	0.0099	n.d.	25	0.0033	1.1	979
Piedra Copahue	PC	4	Nov-06	93	961	n.d.	n.d.	n.d.	10	15	6.4	0.055	0.19	7.1	0.061	n.a.	n.d.	9.0	0.0006	0.89	960
Piedra Copahue	PC	4	Mar-12	95	947	n.d.	n.d.	n.d.	8.9	19	10	0.013	0.008	14	0.011	0.0021	n.d.	8.4	0.0007	0.75	970
Piedra Copahue	PC		Mar-13	94	961	n.d.	n.d.	n.d.	8.6	17	6.1	0.016	0.009	6.6	0.012	0.0019	n.d.	8.6	0.0008	0.88	973
Piedra Copahue	PC		Mar-14	93	962	n.d.	n.d.	n.d.	9.1	21	7.9	0.023	0.007	7.5	0.011	0.0026	n.d.	9.1	0.0007	0.91	982
Piedra Copahue	PC		Mar-15	95	930	n.d.	n.d.	n.d.	11	26	14	0.026	0.011	19	0.021	0.0029	n.d.	9.5	0.0008	0.96	979
Piedra Copahue	PC		Feb-16	95	919	n.d.	n.d.	n.d.	13	23	10	0.021	0.013	35	0.016	0.0088	n.d.	15	0.0028	0.82	983
Anfiteatro	AF	4	Feb-07	92	864	n.d.	n.d.	n.d.	90	13	27	0.061	0.23	5.0	0.089	n.a.	n.d.	2.5	0.0003	0.30	962
Anfiteatro	AF	4	Mar-12	92	870	n.d.	n.d.	n.d.	24	35	55	0.025	0.013	26	0.030	0.0029	n.d.	47	0.0036	5.1	983
Anfiteatro	AF		Mar-13	93	889	n.d.	n.d.	n.d.	18	27	53	0.31	0.015	13	0.018	0.0018	n.d.	10	0.0029	4.8	976
Anfiteatro	AF		Mar-14	93	892	n.d.	n.d.	n.d.	23	31	36	0.13	0.019	18	0.015	0.0026	n.d.	11	0.0024	4.1	985

Sample	Acronym	Acronym Reference Date	Date	T°C	T°C CO ₂	HF	HCI	SO_2	H_2S	\mathbf{N}_2	CH4	Ar	02	H_2	He	co	C_2H_4	C_2H_6	C ₃ H ₆	C_3H_8	Steam
Anfiteatro	AF		Mar-15 94	94	861	n.d.	n.d.	n.d.	20	36	49	0.08	0.013	33	0.018	0.0048	n.d.	13	0.0028	4.5	986
Anfiteatro	AF		Feb-16	92	866	n.d.	n.d.	n.d.	25	34	16	0.09	0.011	59	0.016	0.018	n.d.	29	0.0064	2.1	986
Chancho Co	CC	4	Feb-07	86	911	n.d.	n.d.	n.d.	68	3.8	0.68	0.018	0.23	16	0.035	n.a.	n.d.	7.5	0.0004	0.84	984
Chancho Co	CC	4	Mar-12	95	789	n.d.	n.d.	n.d.	155	14	2.1	0.035	0.011	40	0.016	0.0023	n.d.	2.6	0.0003	0.21	992
Chancho Co	CC		Dec-12	92	814	n.d.	n.d.	n.d.	138	13	2.3	0.17	0.008	33	0.012	0.0019	n.d.	6.9	0.0004	0.56	988
Chancho Co	CC		Jan-13	94	760	n.d.	n.d.	n.d.	169	8.6	7.2	0.15	0.015	95	0.015	0.0058	n.d.	7.8	0.0009	0.61	991
Chancho Co	CC		Mar-14	95	742	n.d.	n.d.	n.d.	151	9.5	6.6	0.19	0.021	91	0.011	0.0048	n.d.	6.5	0.0007	1.1	987
Chancho Co	CC		Mar-15	95	748	n.d.	n.d.	n.d.	156	11	7.1	0.11	0.065	LL	0.014	0.0061	n.d.	6.7	0.0008	1.3	985
Chancho Co	CC		Feb-16	76	737	n.d.	n.d.	n.d.	140	13	5.1	0.09	0.019	105	0.015	0.043	n.d.	16	0.0036	1.1	983
Cristina 1	CR1		Mar-13	431	740	2.9	5.1	165	8.7	16	0.11	0.021	0.005	91	0.023	0.051	0.13	0.56	0.008	0.011	902
Cristina 2	CR2		Mar-13	102	820	2.6	3.8	120	8.9	14	0.19	0.026	0.008	56	0.019	0.033	0.11	0.66	0.006	0.012	936

 Table 1 (continued)

(Capaccioni et al. 2016) and empirical (Tamburello et al. 2015; de Moor et al. 2016b) investigations, as well as theoretical computations (Symonds et al. 2001), have demonstrated that magmatic gas dissolution in liquid water can be negligible at low pH and relatively high temperature. Hence, the occurrence of significant compositional changes in the hydrothermal emissions of Copahue volcano during the new eruptive cycle which started in 2012 was taken seriously. On the contrary, no clues of the typical magmatic gases SO₂, HCl and HF were detected in the investigated distal fumaroles during the whole observation period, whereas these gases were found in significant concentrations in the fluids at the two summit crater fumaroles (Table 1). Therefore, the geothermal aquifers surrounding the volcano efficiently scrubbed these highly soluble gases after the 2012 eruptive activity, despite the fact that the magmatic input at the root of the aquifer was likely increasing (Tassi et al. 2016). The outlet temperatures of most fumaroles were at the boiling point of water at the vent altitude, being likely controlled by the presence of a liquid phase at shallow depth (~ 90 °C; Table 1). Thus, it is not surprising that they remained constant during the observation period, apart from those at LMM, where a superheated steam (T > 100 °C) showed a significant temperature increase (20-25 °C; Table 1) in March-December 2012.

The concentrations of H₂S in all the investigated hydrothermal fluid discharges did not follow any specific temporal trend (Fig. 2a). Similarly, CH₄ showed wide variations randomly distributed in time (Fig. 2b). The N₂/Ar ratios oscillated around that measured at the CR1 fumarole, mostly ranging between those of air and a deep component (DC) (Fig. 2c), the latter having an N_2 /Ar ratio up to 3500 (Agusto et al. 2013) likely related to sediments derived from the slab rather than continental crust (Tardani et al. 2016). Thus, no evident effects related to the last eruptive phase were shown by the behavior of H₂S, CH₄, N₂ and Ar, at least with this sampling frequency. Molecular hydrogen is commonly used as a geo-indicator to estimate temperature and redox conditions regulating the chemistry of hydrothermal fluids, since it reacts rapidly to physical and chemical changes (e.g. Giggenbach 1987). Prior to February 2007, the H₂ concentrations were relatively low and constant, whereas in March 2012, a first concentration peak was observed, followed in January 2013 by a second peak, which only occurred at CC, LM and PC. The March 2013-March 2014 period was marked by decreasing trends, whereas a new phase of increase was recorded during the last two years of observation (Fig. 2d). A relationship between the H_2 temporal patterns and the main phreatic and phreatomagmatic eruptive events is clearly shown, especially considering that the gas discharges displayed synchronous variations. The January 2013 peak and the sharp increase during the last observation period were also found

Table 2 δ^{13} C-CO₂ (‰ vs. V-PDB), *Rc/Ra*, ⁴He/²⁰Ne, δ^{18} O-H₂O (‰ vs. V-SMOW) and δ D-H₂O (‰ vs. V-SMOW) values, and CO₂/³He of the CCVC gas discharges. References (1)–(5) as in Table 1; (6) Varekamp et al. (2006)

Acronym	Reference	Date	δ^{13} C-CO ₂	Rc/Ra	⁴ He/ ²⁰ Ne	$CO_2/{}^{3}He(\times 10^{9})$	$\delta^{18}\text{O-H}_2\text{O}$	δD-H ₂ O
LM	2	1986					- 10.6	- 85
LM	6	2003	- 8.3	7.50	729			
LM	4	Nov-06	- 8.2	7.72	1898	3.3		
LM	4	Feb-07	- 8.8					
LM	4	Mar-12	- 7.3	7.04	1215	6.2	- 11.2	- 90
LM		Dec-12	- 6.9	6.67	115	8.5		
LM		Jan-13	- 7.1	6.48	333	9.6		
LM		Mar-13	- 7.9	6.43	485	8.9	- 8.6	- 76
LM	5	Jan-14	-8.0	7.52	68	8.2		
LM		Mar-14	- 8.6	7.60	571	7.5	- 10.6	- 87
LM		Mar-15	- 9.6	7.65	1025	5.9	-10.7	- 90
LM		Feb-16	- 9.4	7.71	845	6.2	- 10.3	- 92
LMM	2	1986					- 10.8	- 84
LMM	4	Nov-06	- 7.7	7.94	1679	3.1		
LMM	4	Mar-12	- 7.1	6.97	1330	5.2	- 10.8	- 88
LMM		Mar-13	- 7.8	5.89	354	8.0	-8.8	- 78
LMM		Mar-14	- 8.2	7.67	420	8.1	- 10.5	- 90
LMM		Mar-15	- 8.3	7.71	644	6.7	- 10.8	- 91
LMM		Feb-16	- 8.1	7.68	512	5.7	- 10.6	- 92
PC	6	2003	- 8.3	7.60	1228			
PC	4	Nov-06	- 7.7	7.36	1957	1.5		
PC	4	Mar-12	- 7.0	7.01	1048	8.8	- 11.8	- 91
PC		Mar-13	- 6.7	6.79	505	8.5	- 8.2	- 80
PC		Mar-14	- 7.5	7.74	785	8.1	- 9.8	- 88
PC		Mar-15	- 8.2	7.96	1045	4.0	- 9.9	- 89
PC		Feb-16	- 8.7	7.94	913	5.2	- 9.6	- 86
AF	4	Feb-07	- 6.8	5.01	2445	1.4		
AF	4	Mar-12	- 7.0	4.93	1996	4.3	- 17.9	- 109
AF		Mar-13	- 7.6	4.63	23	7.7	- 8.6	- 79
AF		Mar-14	- 8.5	5.12	216	8.4	- 8.5	- 80
AF		Mar-15	- 8.5	5.09	254	6.8	- 8.7	- 82
AF		Feb-16	-8.8	5.16	217	7.5	- 8.5	- 83
CC	4	Feb-07	- 8.3	5.02	914	3.8		
CC	4	Mar-12	- 8.1	4.04	1058	8.5	- 8.6	- 89
CC		Dec-12	- 7.8	4.21	144	12		
CC		Jan-13	- 8.3	4.07	84	9.0		
CC		Mar-14	- 8.8	5.35	75	9.1	- 8.3	- 78
CC		Mar-15	- 9.3	5.12	217	7.5	- 8.2	- 75
CC		Feb-16	- 9.2	5.22	186	6.8	- 8.4	- 77
CR1		Mar-13	- 9.6	7.75	458	3.0	- 3.5	- 66
CR2		Mar-13	- 9.5				- 3.6	- 70

in the CO concentrations (Fig. 2e) and the C_3H_6/C_3H_8 ratios (Fig. 2f). These data suggest that the magmatic pulses likely causing the volcanic events at Copahue volcano also affected the hydrothermal system, increasing its temperatures and, consequently, enhancing the formation of both H_2 , CO and C_3H_6 . The significant temperature increase that affected the hydrothermal reservoir in 2015–2016 is clearly shown in Fig. 3, where the redox independent gas geothermometer proposed by Chiodini and Marini (1998) is reported.

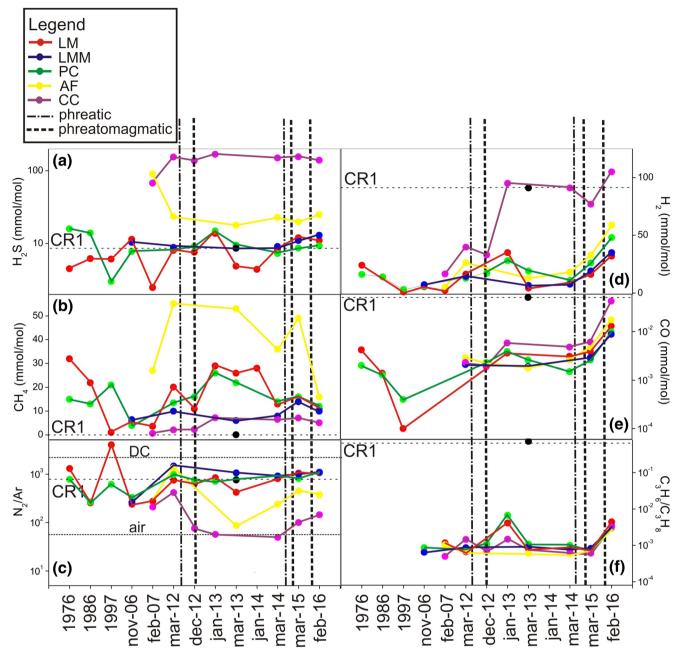


Fig. 2 Time-series of (a) H_2S , (b) CH_4 , (c) N_2/Ar , (d) H_2 , (e) CO, and (f) C_3H_6/C_3H_8 from 1976 to February 2016 of gases from the LM (red line), LMM (blue line), PC (green line), AF (yellow line) and CC (magenta line) fluid discharges. The main phreatic (dash-dotted line) and

Temporal evolution of the δ^{13} C-CO₂, *Rc/Ra*, δ^{18} O-H₂O and δ D-H₂O values

The Copahue fluids are marked by a high He isotopic ratio (Varekamp et al. 2006; Agusto et al. 2013). It is the highest one measured in the SAVZ (Tardani et al. 2016 and references therein) and fits into the MORB field ($8 \pm 1 Ra$; Farley and Neroda 1998; Graham 2002). This peculiar feature is likely related to crustal attenuation and upwelling of mantle asthenosphere caused by the transtensional tectonics that caused the

phreatomagmatic (dash line) eruptions, as well as the CR1 fumarole (black dotted line), are also reported. In (c), the N_2 /Ar ratios of air and a deep component (DC) are also depicted

CCVC pull-apart basin (Folguera et al. 2007). As highlighted by Agusto et al. (2013) and Tassi et al. (2016), the CC and AF gases are characterized by a relatively high degree of crustal contamination, possibly due to shallow fluids permeating through the normal fractures affecting this sector of the volcano (Folguera et al. 2016). In March 2012, i.e. just before the last eruptive cycle, the *Rc/Ra* values progressively decreased in all the investigated fluid discharges. In March 2013, the *Rc/Ra* values were 1–2 units lower than those measured in 2003–2007, whereas since March 2014, synchronous

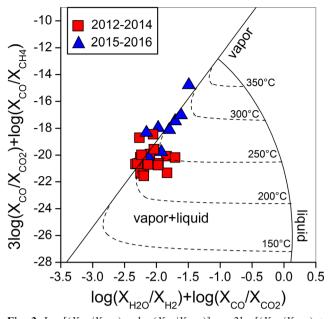


Fig. 3 $\text{Log}[(X_{CO}/X_{CO2}) - \log(X_{H2}/X_{H2O})]$ vs. $3\log[(X_{CO}/X_{CO2}) + \log(X_{CO}/X_{CH4})]$ binary diagram for the CCVC fumaroles. The theoretical values for a single saturated vapor phase (vapor) and single saturated liquid phase (liquid) are shown. Gases are grouped on the basis of their sampling date: 2012–2014 (red squares); 2015–2016 (blue triangles)

increasing trends have been recorded. In February 2016, the Rc/Ra values were comparable to, or even higher than, those measured prior to 2012 and those of CR1 (Fig. 4a). The simultaneous variations of the Rc/Ra values in the studied fumaroles are further evidence for a common deep source of He, notwithstanding the relatively high crustal contamination shown by the gases from CC and AF with respect to the other hydrothermal fluid discharges.

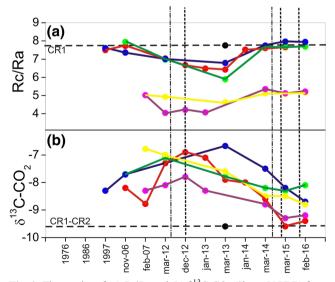


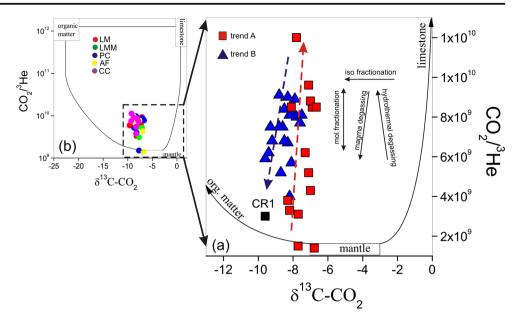
Fig. 4 Time-series of (a) Rc/Ra and (b) δ^{13} C-CO₂ (‰ vs. V-PDB) from 2003 to February 2016 of gases from the CCVC fumaroles. Symbols as in Fig. 2

Contrasting temporal patterns were given by the δ^{13} C-CO₂ values, which in 2012 were significantly less negative than those from 2003 to 2007, apart from at AF where the carbon isotope composition in CO₂ showed a decreasing trend over the whole observation period (2007–2016). From 2013, the δ^{13} C-CO₂ values of the LM, LMM and CC discharges also started to decrease, whereas the PC fumarole showed an increasing trend that lasted until March 2013 and then abruptly decreased. The δ^{13} C-CO₂ values in 2015–2016 were the lowest among those measured for these five fumaroles (Table 2), approaching those of CR1 and CR2 (Fig. 4b).

The Rc/Ra and δ^{13} C-CO₂ temporal patterns (Fig. 4a,b) produced two distinct trends in the $CO_2/{}^{3}He vs. \delta^{13}C-CO_2$ binary diagram (Fig. 5): trend A, including those gases that were marked by significant $CO_2/{}^{3}$ He and δ^{13} C-CO₂ increases in 2006–2013; and trend B, delineated by the gases whose $CO_2/^3$ He and δ^{13} C-CO₂ values, after the period of increase, progressively approached those measured in 2006-2007. An enhanced degassing of the hydrothermal reservoir may theoretically produce an increase in the $CO_2/{}^3$ He ratio, due to the higher CO₂ solubility in water with respect to that of He (CRC 2001). If we assume that this process was related to the eruption occurring at the active crater, then the increase in the reservoir temperature, suggested by the behavior of the gas species most sensitive to temperature changes (Fig. 2), is not surprising. However, an increasing rate of gas release from the hydrothermal reservoir would decrease the gas-liquid ¹³C-¹²C fractionation, leading to a decrease in δ^{13} C-CO₂ values (Deines et al. 1974) towards the CR1-CR2 values, i.e. the opposite of what is observed in 2012-2014 (Fig. 4b). Hence, this process cannot completely explain trend A.

As far as trend B is concerned, an increase in magma degassing causes a preferential loss of CO₂ with respect to He, the latter having a higher solubility in the melt than the former, thus producing gases characterized by a progressive decrease in both the $CO_2/{}^{3}$ He and δ^{13} C-CO₂ values (Javoy et al. 1978; Hilton et al. 1998). This would explain, at a first approximation, the changes of these two parameters which occurred after 2014 (Fig. 5). An enhanced contribution of magmatic gases is consistent with the increase in the Rc/Ravalues observed in 2015-2016 (Fig. 4a), but it seems to be in contradiction with the Rc/Ra decrease observed in 2012-2014, when the last intermittent eruptive cycle, representing the clearest expression of renewed magmatic activity, had already begun. The Rc/Ra evolution unequivocally indicates that in 2012–2014 the hydrothermal fumaroles received a significant contribution of crustal-type fluids. Overall, the occurrence of an anomalous input of ⁴He- and ¹³C-rich fluids to the hydrothermal fumaroles in 2012-2014 is the only hypothesis able to reconcile the variations in Rc/Ra, $CO_2/{}^{3}He$ and $\delta^{13}C$ -CO₂ values, although it is in apparent contradiction with the evolution of the volcanic activity. Differential SAR Interferometry (DInSAR) measurements showed that, after a

Fig. 5 (**a**,**b**) CO₂/³He vs. δ^{13} C-CO₂ (‰ vs. V-PDB) binary diagram of gases from the CCVC fumaroles. Panel **b** is a zoom of panel **a**. Gases collected in 2006– 2013 (trend *A*): red square; gases collected in 2014–2016 (trend *B*): blue triangle. The CR1 fumarole (black square) is also reported. Boxes and mixing curves (in panel **a**) between the three endmembers (organic sediment, limestone and MORB) are from Sano and Marty (1995)



long deflation phase (Velez et al. 2011), Copahue volcano underwent an inflation process from February 2011 to at least January 2013 (no data are available afterward) (Veléz et al. 2016). This inflation, which was interpreted as being related to magma input into the deep (7.5–9 km) chamber that likely promoted the 2012-2016 eruptive cycle, may have caused tensional stress at the surface, favoring the permeation of shallow fluids that affected the chemistry of the uprising hydrothermal gases. The mixing between deep and shallow components, a process that is also responsible for the relatively low Rc/Ra values characterizing the CC and AF fumaroles (Table 2), almost completely masked the increasing heat released from the magmatic system to the root of the hydrothermal reservoir until 2014. The H₂, CO and C₃H₆/C₃H₈ peaks shown in January 2013 by most fumaroles (Fig. 2e-f) were likely caused by the enhanced interaction between the magmatic and hydrothermal fluids. The opposite trends shown by the Rc/Ra and δ^{13} C-CO₂ values in 2012–2013 (Fig. 4a,b, respectively) may have an alternative explanation. Under the stress related to the intense seismic activity that occurred prior to the 2012 eruptive events, crustal (⁴He- and ¹³C-rich) fluids may have been released, a process that is commonly considered as being precursory to earthquakes (Khilyuk et al. 2000). Strong variations in He isotopes associated with seismic crisis have also been described by other authors (e.g. Padron et al. 2013). The inversion of the *Rc/Ra* and δ^{13} C-CO₂ trends which occurred in 2013 (Fig. 4a,b) indicates that the crustal gas contribution (from permeation of shallow fluids and/or gases released from crustal rocks) started to decrease, allowing the progressive increase of the magmatic gas fraction. Accordingly, the Rc/Ra, CO₂/³He and δ^{13} C-CO₂ values characterizing the hydrothermal discharges in 2006-2007, as well as the CR1 fumarole, were almost completely restored in 2015–2016. Further evidence for this process is the remarkable increase in H₂ and CO concentrations (Fig. 2e,d) and C3H6/C3H8 ratios (Fig. 2f). It is worth noting that CR1 showed a δ^{13} C-CO₂ value which was more negative than the lower limit of the range of values for both MORB and high-temperature volcanic gases (-6.5 ± 2.2 and $-5.5 \pm 2.2\%$ vs. V-PDB, respectively; e.g. Sano and Marty 1995; Sano and Williams 1996), and a $CO_2/{}^{3}$ He ratio lower than that related to global arc-like values $(1.5 \pm 1.1 \times 1010)$; e.g. Sano and Williams 1996). Hence, assuming that the chemical features of this high-temperature fumarole (sampled in March 2013) approached those of the magmatic fluids, the latter seem to have suffered a certain degree of CO₂-He and ¹³C-¹²C fractionation processes, which were likely produced by the enhanced magma degassing related to the 2012 eruptive events. Evidences of magmatic fluids in the hydrothermal discharges were also traced by the δ^{18} O-H₂O and δ D-H₂O values. When only those of the March 2012 samples were considered, they provided an indication of ¹⁸O enrichment caused by water-rock interaction processes (Fig. 6; Agusto et al. 2013). On the contrary, according to the new complete dataset, a convergence with the CR1 and CR2 samples was shown (Fig. 6), suggesting a significant contribution of andesitic water (Taran et al. 1989; Giggenbach 1992).

Conclusions

The geochemical monitoring carried out over the last 40 years at Copahue volcano, although discontinuous, has provided useful insights into the mechanisms regulating the chemical

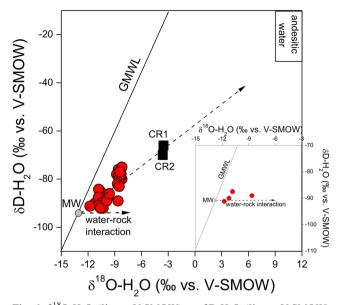


Fig. 6 δ^{18} O-H₂O (‰ vs. V-SMOW) vs. δ D-H₂O (‰ vs. V-SMOW) binary diagram of gases from the CCVC fumaroles. In the inset, the binary diagram is reported at different *x* and *y* scales. The isotopic compositions of local meteoric water (MW), the Global Meteoric Water Line (GMWL; Craig 1961) and the "andesitic" water field (Taran et al. 1989; Giggenbach 1992) are also reported. Hydrothermal fumaroles: red circle; active crater fumarole: black square

and isotopic changes that affected the hydrothermal fluid discharges during the 2012–2016 eruptive cycle. The H₂ and CO concentrations, as well as the ratio between the C3 alkenealkane pair, showed occasional peaks (in March 2012, January 2013 and in 2015–2016; Fig. 2), indicating episodes of disturbance of the fumarolic fluids in correspondence with the main volcanic events that occurred at the active crater. The behavior of these geochemical parameters revealed that the enhanced input of hot magmatic fluids at the root of the hydrothermal reservoir related to the volcanic activity produced, albeit with a time delay, a detectable signal at the surface. On the contrary, the temporal evolution of the δ^{13} C-CO₂, *Rc/Ra* and CO2/3He values in 2006-2014 was unexpected, since they clearly indicated a significant decrease in the magmatic fluid contribution over a period in which several eruptions occurred. The two most reliable hypotheses for this behavior are based on (a) the effect of the inflation process, affecting the volcano since early 2011, on ground permeability that favored the mixing between the uprising hydrothermal fluids and a shallow ¹³C-rich and ³He-poor end-member and (b) an enhanced release of gases from crustal rocks associated with the earthquakes that occurred prior to the 2012 eruptive activity. Only since 2013 have the δ^{13} C-CO₂ and *Rc/Ra* values evolved towards a progressive increase in the magmatic component.

Overall, the interpretation of this large time-series of geochemical data highlighted that the hydrothermal fluids discharged from the peripheral vents did not show any precursory signals of the new incoming eruptive phase; heat and magmatic gases were efficiently buffered by the hydrothermal reservoir. Hence, the compositional changes of the fluids discharged at the surface were smoothed and delayed. Moreover, a frequent (e.g. monthly) sampling of these gases is difficult due to the adverse meteorological conditions that characterize the area for most part of the year. To forecast rejuvenation of volcanic activity, which in this volcano typically occurs at the active crater (Caselli et al. 2016b), the installation of a geophysical array (Caselli et al. 2016c and references therein) combined with ground deformation measurements from satellites (Veléz et al. 2016), as well as geochemical monitoring of the stream (namely Rio Agrio) fed by the hot and hyperacidic springs located close to the rim of the active crater (Agusto and Varekamp 2016), seems the most appropriate approach. A MultiGAS instrument could be placed at the summit crater rim to provide medium-to-longterm (on the scale of months) data series of selected geochemical parameters (e.g. CO₂/Ar, H₂/Ar), possibly able to detect changes in the volcanic gas emission rate from the central degassing system even during quiescent periods. On the other hand, the relative stability of this hydrothermal system, persisting even during a volcanic crisis, represents a fundamental advantage for geothermal exploitation, which may cause renewed interest in this resource following the first promising attempts carried out over the past few decades. However, the increase in H₂ and CO concentrations and that of the C₃H₆/C₃H₈ ratio observed after March 2015 deserves particular attention. A new period of unrest at Copahue volcano followed by new volcanic events may indeed significantly modify the geochemical composition of the peripheral gas emissions since in this case the presence of magmatic fluids in these discharges cannot be completely ruled out. Consequently, periodic geochemical monitoring should be performed independent of further developments of geothermal projects.

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